THIELAVIN A AND B, NEW INHIBITORS OF PROSTAGLANDIN BIOSYNTHESIS PRODUCED BY *THIELAVIA TERRICOLA*

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Two potent inhibitors of prostaglandin biosynthesis, thielavin A ($C_{31}H_{34}O_{10}$) and B ($C_{29}H_{30}O_{10}$), were isolated from cultures of *Thielavia terricola*. Both of these compounds were shown to be structurally related to depsides, thus consisting of three hydroxybenzoic acid groups. Concentrations required for 50% inhibition of the conversion of ¹⁴C-arachidonic acid into prostaglandins $F_{2\alpha}$ plus E_2 by microsomes of ram seminal vesicles were 12 μ M for thielavin A and 9 μ M for thielavin B, respectively. Of the enzymatic steps involved in prostaglandin synthesis, thielavin A specifically inhibited the conversion of arachidonic acid into prostaglandin H_2 , while prostaglandin E_2 synthesis from the endoperoxide was the most sensitive to thielavin B.

Thromboxane A_2 synthesis from prostaglandin H_2 in bovine platelet microsomes were inhibited by 50% at concentrations of 150 and 350 μ M of thielavin A and B, respectively.

Thielavin B was significantly effective on carrageenan-induced oedema of rats when administered intravenously but not on oral administration. The anti-inflammatory activity was not detectable with thielavin A either on intravenous injection or on oral administration.

Non-steroidal anti-inflammatory agents like aspirin and indomethacin are known to specifically inhibit prostaglandin synthetase.^{1~3)}

During a search for specific inhibitors of prostaglandin synthesis of microbial origin, xanthocillin X monomethyl ether as a potent inhibitor was isolated from cultures of *Dichotomomyces cejpii*.⁴⁾

Further study has resulted in the isolation of two active compounds (thielavin A and B) from cultures of *Thielavia terricola*.

The present paper deals with identification of the producing organisms and production and isolation of thielavin A and B as well as their physicochemical properties and biological activities. The results of studies dealing with the structural elucidation of these inhibitors will be reported in a subsequent paper.

Materials and Methods

Chemicals

 1^{-14} C Arachidonic acid (61.0 Ci/mole) was obtained from New England Nuclear and unlabeled arachidonic acid (grade 1), prostaglandin $F_{2\alpha}$ and E_2 , reduced glutathione and L-tryptophan from Sigma, bovine hemoglobin from Tokyo Kasei, imidazole and silica gel (Wako gel C-200) from Wako Pure Chemical Industries. Indomethacin was kindly given by Dr. E. MISAKA of this laboratory.

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Culture Medium

Fermentation medium for the production of thielavin A and B contained 10.0% soluble starch, 1.5% Pharmamedia (Traders Protein), 2.0% corn steep liquor (Corn Products Co.), 1.0% meat extract, and 0.02% Disfoam CB 442 (Nippon Yushi Co.) an anti-foam agent.

Enzyme Preparations and Enzyme Assays

The microsomal fraction of ram seminal vesicular glands was obtained by the method of NUGTEREN et $al.^{5}$ and used to assay the synthesis of prostaglandins as described previously.⁴⁾ The microsomal fraction of bovine platelets was prepared according to the method of YOSHIMOTO et $al.^{6)}$ and used to assay thromboxane synthesis from ¹⁴C-prostaglandin endoperoxide (H₂), prepared from ¹⁴C-arachidonic acid according to the method of UBATUBA et $al.^{7)}$

Anti-inflammatory Activity

Anti-inflammatory effects on carrageenan-induced oedema of the rat foot pad were determined using male Wistar-Imamichi rats as described previously.⁴⁾

Results and Discussion

Identification of Producing Fungus

The fungus SANK 15876, which produces thielavin A and B, was freshly isolated from a soil sample collected in Thailand.

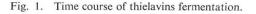
The mycological characteristics of this fungus are as follows: Colonies on WEITZMAN and SILVA-HUTNER's agar⁸⁾ are attaining a diameter of 6 to 7 cm in 10 days at 26°C, which are white and cottony with thin aerial mycelia, and afterwards produce perithecia abundantly as grayish dots. The reverse is uncolored. Growth at 37°C was nearly the same as at 26°C, but perithecial production was less than that of 26°C. The perithecia are superficial, spherical, glabrous, $50 \sim 220 \mu$ in diameter and gray to nearly black at maturity. The cells of peridium are membranaceous to pseudoparenchymatous, thin and semi-transparent. The asci are ovate, 8-spored, $20 \sim 26 \times 15 \sim 18 \mu$ and evanescent. The ascospores

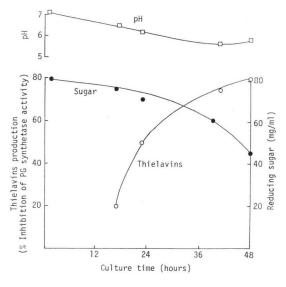
are ellipsoid, $12 \sim 17 \times 7 \sim 8 \mu$, smooth, at first hyaline, then becoming dark olive to olive brown, with a distinct apical germ pore at one end. No conidial structures are observed.

From the mycological characteristics described above, the fungus was identified as *Thielavia terricola* (Gilman et Abbott) Emons.^{9~12)}

Fermentation

Thielavia terricola SANK 15876 was aerobically grown at 26°C in a 30-liter fermentor containing 18 liter of culture medium, inoculated with 2 liters of seed culture, under an aeration rate of 18 liters/minute and agitation at 150 r.p.m.. Under these conditions, production of the active principles reached maximal levels after 48 hours (Fig. 1.).





Isolation of Inhibitors

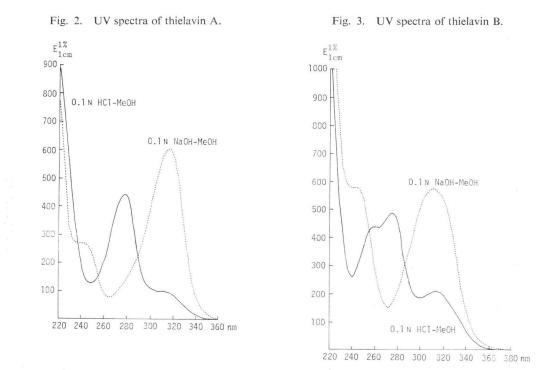
The inhibitory principles were found to be located in the mycelia but not in the culture filtrate.

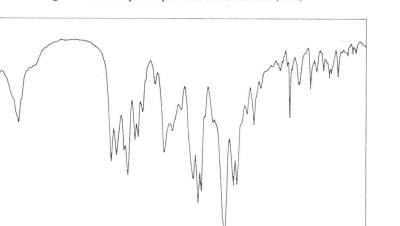
After 48 hours of growth mycelia of *Thielavia terricola* SANK 15876 were collected by filtration from 2 cultures of 30-liter fermentor, and the mycelial mat obtained (3.8 kg wet weight) was extracted with 10 liters of acetone at room temperature for 30 minutes. The mixture was filtered and the resultant filtrate was concentrated *in vacuo* to remove acetone. The aqueous solution obtained (2 liters) was adjusted to pH 2.0 with H_3PO_4 and extracted twice with 1 volume of ethyl acetate. The solvent layers collected were concentrated *in vacuo*, yielding 25 g of oily fraction. The oily residue was applied to a column of silica gel C-200 (3 × 60 cm) equilibrated with benzene - *n*-hexane (1: 1), and after washing with 1 liter of benzene, the column was developed with 1% ethyl acetate in benzene. Fractions containing thielavin B were pooled and concentrated *in vacuo* to dryness, giving 2 g of white powder. The powder resulting was dissolved in *n*-hexane from which solution thielavin B was obtained as crystals (1.2 g).

After elution of thielavin B, the column was developed with ethyl acetate - benzene (1:1), and the active fractions were pooled and concentrated *in vacuo* to dryness, giving 1 g of yellowish powder. This powder was rechromatographed on a silica gel column $(2.6 \times 50 \text{ cm})$; the column was developed as described above. Under these conditions, one major active fraction and one minor active fraction were eluted by development with ethyl acetate - benzene (1:1). To the major fraction was added *n*-hexane and the resultant crystals (thielavin A) were isolated and dried (0.51 g).

Physico-chemical Properties of the Thielavins

Thielavin A and B were obtained as colorless needle crystals from acetone. Melting points were $235 \sim 236^{\circ}$ C and 250° C for thielavin A and B, respectively, and both substances showed no optical rotation. These compounds were soluble in methanol, ethanol, butanol, acetone, ethyl acetate, butyl acetate, ethyl ether, chloroform, dioxane, dimethyl sulfoxide, and pyridine but almost insoluble in water, benzene, and *n*-hexane.







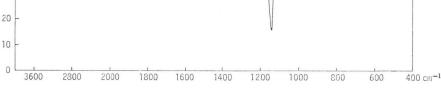
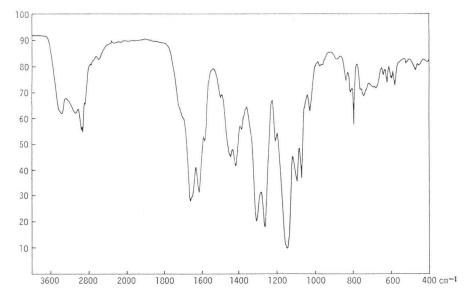


Fig. 5. IR absorption spectrum of thielavin B (KBr).

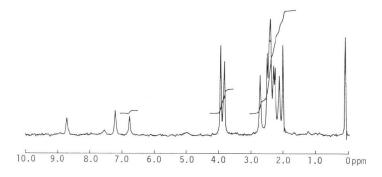


Both thielavin A and B were positive to ammonia phosphomolybdate, potassium ferricyanide - ferric chloride, and hydroxamic acid - ferric chloride, KMnO₄, but negative to ninhydrin, FEHLING and MOLISH reactions. The UV spectra of thielavin A and B are shown in Figs. 2 and 3, respectively. Thielavin A had absorption maxima at 277 nm ($E_{1em}^{1\%}$ 445) and 310 nm ($E_{1em}^{1\%}$ 100) in methanol or 0.1 N HCl - methanol, and at 241 nm ($E_{1em}^{1\%}$ 270) and 315 nm ($E_{1em}^{1\%}$ 605) in 0.1 N NaOH - methanol.

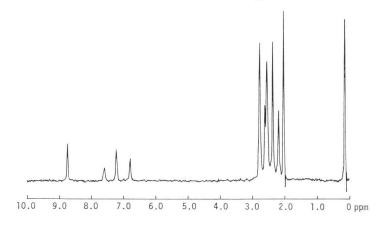
Thielavin B showed absorption maxima at 260 nm ($E_{1em}^{1\%}$ 440), 274 nm ($E_{1em}^{1\%}$ 490) and 314 nm ($E_{1em}^{1\%}$ 210) in methanol or 0.1 N HCl-methanol, and at 243 nm ($E_{1em}^{1\%}$ 580) and 310 nm ($E_{1em}^{1\%}$ 570) in 0.1 N

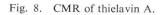
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Fig. 6. PMR of thielavin A (in pyridine).









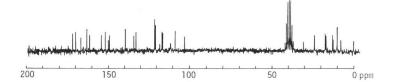


Fig. 9. CMR of thielavin B.

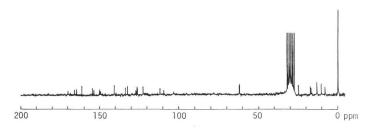
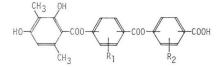


Fig. 10. Partial structure of thielavin A and B.



Thielavin A: $R_1 = R_2 = CH_3 \times 3$, $OCH_3 \times 1$ Thielavin B: $R_1 = R_2 = CH_3 \times 3$, $OH \times 1$

Table 1. Concentration required for 50% inhibition (ID_{50}) in (I) the conversion of arachidonic acid (AA) into prostaglandin H₂ (PG H₂), (II) the conversion of prostaglandin H₂ (PG H₂) into prostaglandin E₂ (PG E₂), and (III) the conversion of PG H₂ into thromboxane A₂.

Inhibitor	$ID_{50}(\mu M)$		
	(I)	(II)	(III)
Thielavin A	10	40	150
Thielavin B	40	9	350
Indomethacin	30	130	
Imidazole			200

NaOH - methanol. These spectrum changes with pH were similar to those obtained with salicylic acid. The IR absorption spectra of thielavin A and B in KBr disk are shown in Figs. 4 and 5, respectively, indicating the presence of carboxylic acid and benzene moieties. The PMR of these compounds are shown in Figs. 6 and 7, suggesting the presence of aromatic proton due to the signal at 6.8 ppm and 8 methyl residues between $2.1 \sim 2.8$ ppm. The data further indicated that thielavin A had two methoxy groups due to the signals at 3.9 and 4.1 ppm. The CMR of thielavin A and B are shown in Figs. 8 and 9, which suggested the presence of 31 carbons in thielavin A and 29 carbons in thielavin B, respectively. Elementary analysis of thielavin A and B was as follows:

Thielavin A:	Found (%)	C: 65.88, H: 6.40
	Calcd. for C ₃₁ H ₃₄ O ₁₀	C: 65.71, H: 6.05
Thielavin B:	Found (%)	C: 64.03, H: 6.29
	Calcd. for $C_{29}H_{30}O_{10}$	C: 64.68, H: 5.61

On the basis of the data presented, the structure of thielavin A and B could be represented as shown in Fig. 10. Further details of structural elucidation will be reported in a subsequent paper.

Biological Activity

Prostaglandin Synthetase Inhibition

Concentrations required for 50% inhibition (ID₅₀) of the conversion of arachidonic acid into prostaglandin H₂ (I), the conversion of prostaglandin H₂ into prostaglandin E₂ (II) and thromboxane A₂ synthetase (III) are summarized in Table 1.

As indicated, thielavin A most strongly inhibited the conversion of arachidonic acid into PG H₂ in the reactions tested. On the other hand, thielavin B specifically inhibited the step involving the synthesis of prostaglandin E₂ from prostaglandin H₂. The inhibitory activity of thielavin A to the conversion I was far stronger than indomethacin, a drug known as a specific inhibitor of reaction I.¹⁸⁾ Thielavin A and B inhibited the thromboxane A₂ synthesis in bovine platelet microsomes by 50% at a concentration of 150 μ M and 350 μ M, respectively, values that were comparable to that for imidazole (200 μ M), a specific inhibitor of III.¹⁴⁾

Anti-inflammatory Activity

Both thielavin A and B showed no significant anti-inflammatory effects on carrageenan-induced oedema in rat foot pad when given orally at a dose 50 mg/kg. When given intravenously, however, thielavin B inhibited this system by 70% at a dose of 5 mg/kg while the anti-inflammatory activity was not significant for thielavin A even on intravenous administration at 5 mg/kg.

Discussion

The present experiments reveals that two novel depsides thielavin A and B are potent inhibitors of prostaglandin synthesis from arachidonic acid. Although structure of these compounds is closely related to each other (Fig. 10), their mode of action is distinct from each other. Thus, thielavin A specifically inhibits the conversion of arachidonic acid into prostaglandin H_2 while thielavin B blocks rather specifically the conversion of the endoperoxide into prostaglandin E_2 . Both compounds might be used as a research tool for studying the regulation of and the mechanism for the synthesis of prostaglandin synthesis.

Differences between thielavins A and B are further seen in their effectiveness on carrageenan-induced oedema on the rat foot pad. Thus thielavin B is effective on intravenous administration while no anti-inflammatory activity can be detectable with thielavin A. At present, however, thielavin B is not useful as an anti-inflammatory agent, since it is not active on oral administration. To improve the activity on oral administration, thielavin B should be modified chemically.

Acknowledgments

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